

Zinc Coordination Polymers Containing Isomeric Forms of p-(Thiazolyl)benzoic Acid: Blue-Emitting Materials with a Solvatochromic Response to Water

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Abstract

© 2017 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim Two coordination polymers of assorted dimensionality (1D, 2D) have been prepared, namely $[\text{Zn}_3(\text{L}2\text{Th})_4(\text{OH})_2 \cdot 2(\text{HL}2\text{Th})] \infty$ (1) and $[\text{Zn}(\text{L}5\text{Th})(\text{OAc})] \infty$ (2), starting from Zn II salts and the isomeric forms of the organic linker p-(thiazolyl)benzoic acid: p-(2-thiazolyl)benzoic acid (HL 2Th) and p-(5-thiazolyl)benzoic acid (HL 5Th). The isomers have been prepared ad hoc, following straightforward Pd-catalyzed C–C coupling reaction protocols. In 1, the deprotonated ligand is coordinated through its carboxylate group only, with dangling thiazole groups. The $-\text{COO}^-$ units are bridging adjacent metal centers, thus creating a 1D chain. The Zn_3 cluster is made of one six-coordinate (Oh) and two four-coordinate (Td) Zn II ions; triple-bridging μ_3 -OH groups are balancing the overall positive charge. The structure of 2 is instead made of Zn_2 (carboxylate) 4 “paddle-wheel” dimers as the constituting inorganic node. The octahedral metal coordination sphere includes two μ -(κ -COO) benzoate spacers, two μ -(κ -COO) acetate ions, the thiazole N atoms coming from adjacent building blocks, and a weak $\text{Zn} \cdots \text{Zn}$ axial interaction. The resulting final assembly is two-dimensional (2D), where p-(5-thiazolyl)benzoate adopts a genuine μ -[$\kappa(\text{COO})\text{:}\kappa(\text{N})$] bridging coordination mode. The luminescent properties of both polymers have been analyzed in the solid state; they feature ligand-centered emissions at $\lambda = 434$ nm (1) and $\lambda = 427$ nm (2). These electronic transitions fall in the visible region, giving the samples a characteristic blue color under an ordinary UV lamp (excitation at $\lambda = 254$ nm). The theoretical analysis of the electronic features of the ligands and related molecular orbitals reveals that the observed transitions are mainly of $\pi \rightarrow \pi^*$ nature, involving π orbitals delocalized on both aromatic cycles. A significant (reversible) blueshift of the emission maximum of ca. 60 nm, from the visible to the UV region, has been observed for 1 when suspended in water.

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Keywords

Coordination polymers, Luminescence, Solvatochromism, Thiazole, Zinc

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